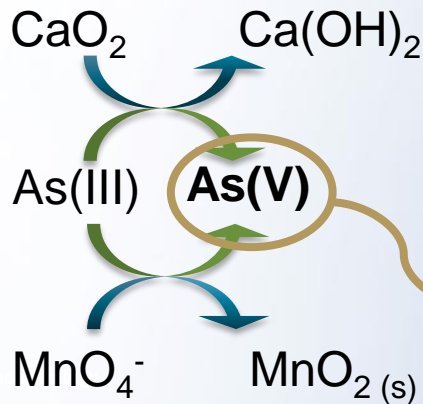


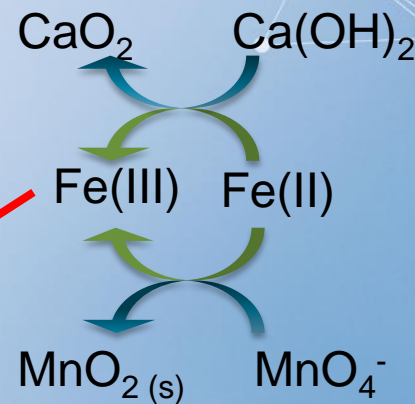
Remedial Technologies to Address CCR Constituents in Groundwater

Environmental Show of the South – May 2018

Arsenic Oxidation



Iron Oxidation



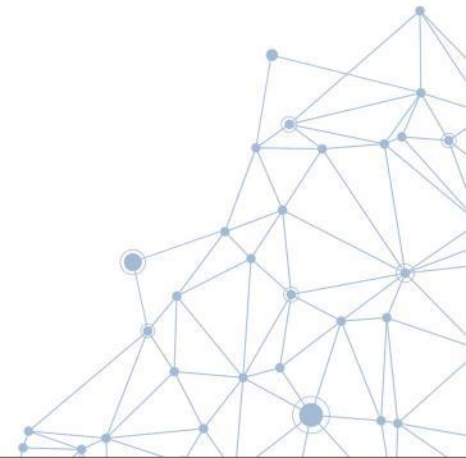
*Sorption,
Coprecipitation*

Oxide Mineral

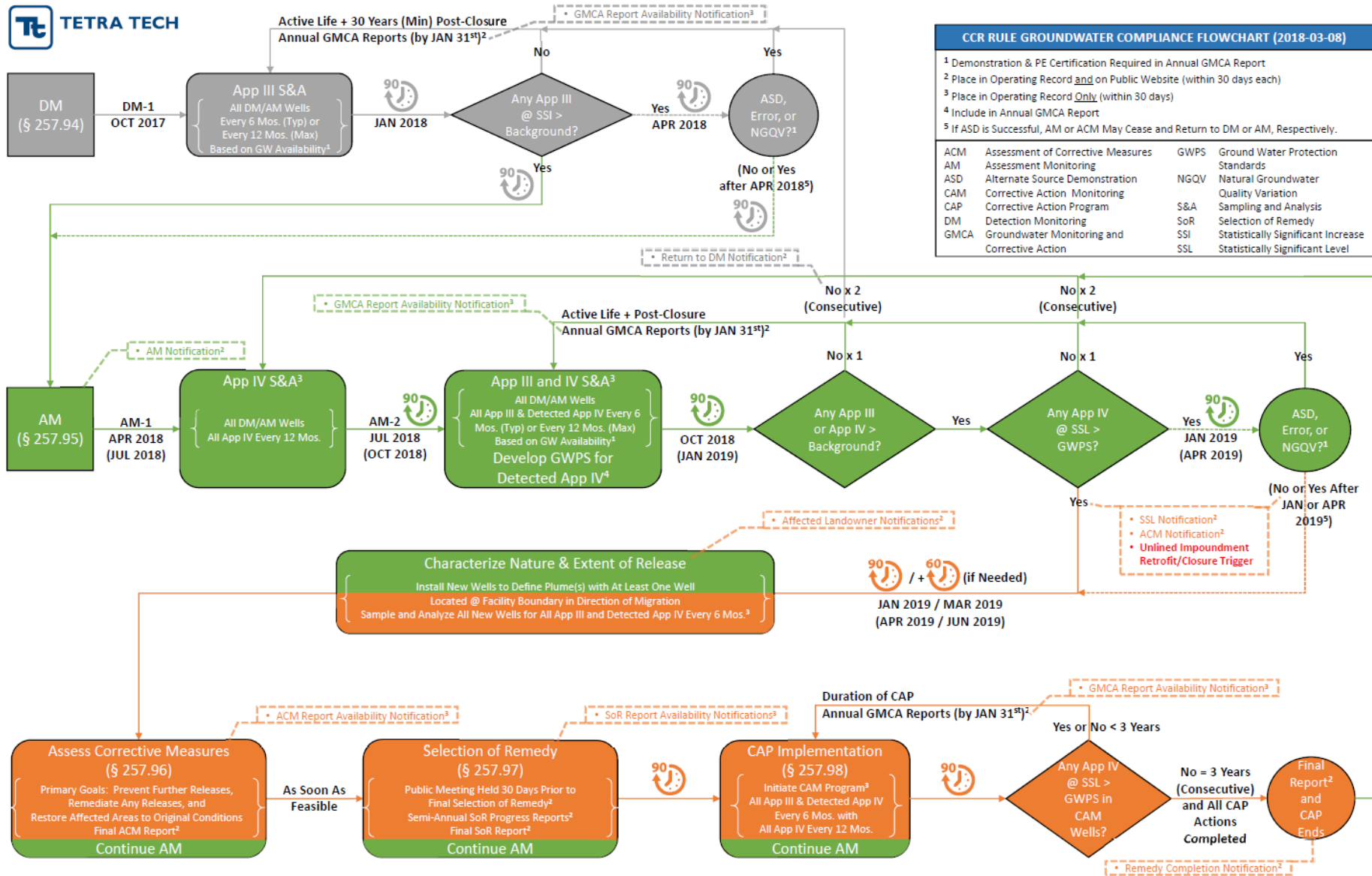
Agenda

- **Topics to be Covered**

- Overview of CCR Rule Groundwater Monitoring Program and Timeline
 - Baseline Monitoring
 - Detection Monitoring
 - Assessment Monitoring
- Corrective Measures Assessment
- Selection of Remedy
- Remediation and Timeline



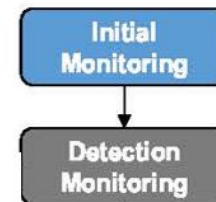
CCR Rule Groundwater Monitoring



Initial (Baseline) Monitoring

- **Minimum of 8 independent samples per well or spring required (background/upgradient and downgradient)**
- **Analyze for all Appendix III and Appendix IV parameters**
- **Sufficient number of samples obtained and analyzed for all four CCR units**

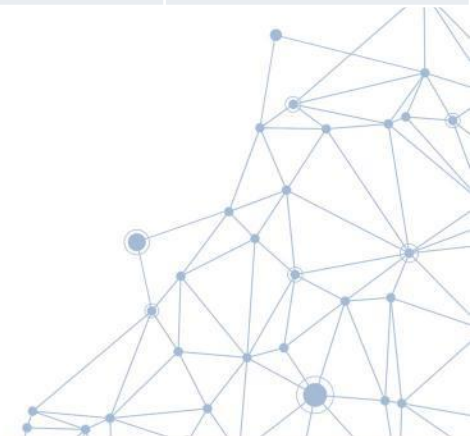
Appendix III	Appendix IV	
Boron	Antimony	Lead
Calcium	Arsenic	Lithium
Chloride	Barium	Mercury
Fluoride	Beryllium	Molybdenum
pH	Cadmium	Radium 226+228
Sulfate	Chromium	Selenium
Total Dissolved Solids	Cobalt	Thallium
	Fluoride	



Statistical Analysis of Data

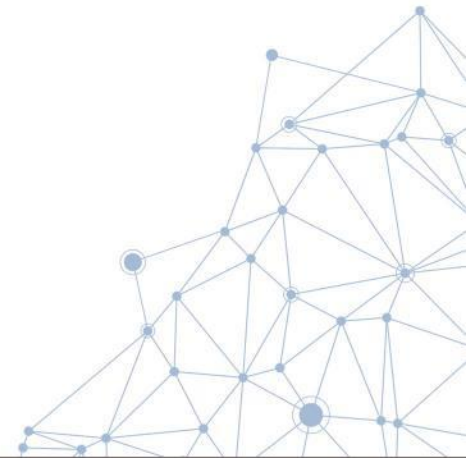
- **Goals of Statistical Analysis:**
 - Determine if statistically significant increase (SSI) greater than background concentrations for each Appendix III parameter for Detection Monitoring
 - Determine if statistically significant level (SSL) greater than MCL or alternative criteria concentrations for each Appendix IV parameter for Assessment Monitoring
 - Determine if Appendix IV parameters in downgradient wells have been below MCL or alternate criteria for 3 consecutive years after Corrective Action Measures implemented
- **Statistical Analysis is Dynamic**

Appendix III	Appendix IV	
Boron	Antimony	Lead
Calcium	Arsenic	Lithium
Chloride	Barium	Mercury
Fluoride	Beryllium	Molybdenum
pH	Cadmium	Radium 226+228
Sulfate	Chromium	Selenium
Total Dissolved Solids	Cobalt	Thallium
	Fluoride	



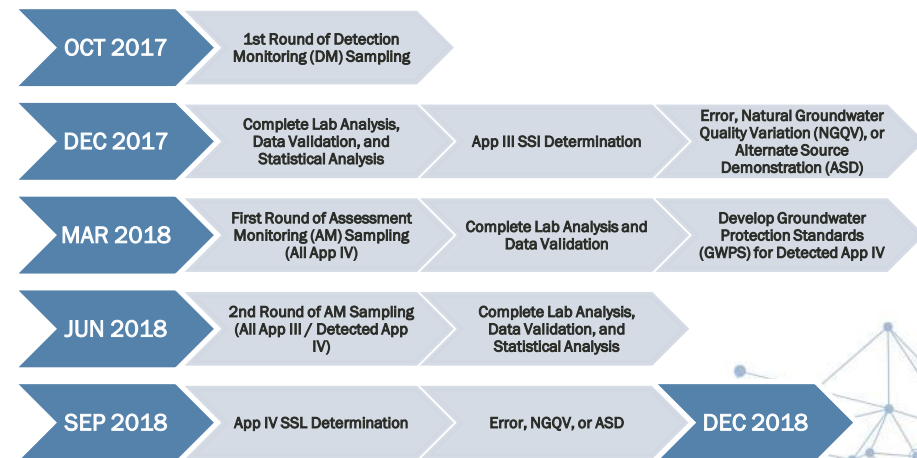
General Characteristics of Sites

- Located near rivers or other major water bodies
- Selection of upgradient and downgradient monitoring well locations
- Sites located on Karst with selection of springs as monitoring locations
- Existing water treatment systems



Next Steps in CCR Compliance Timeline

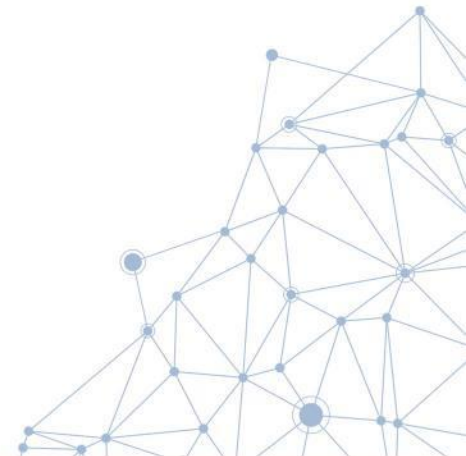
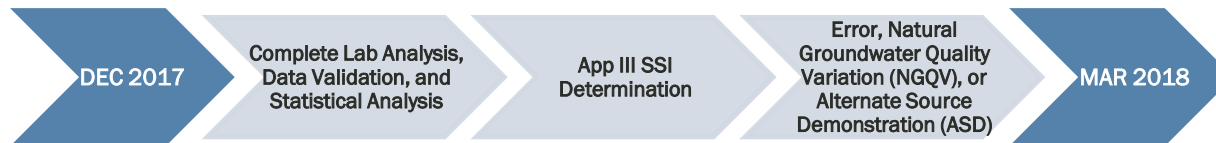
- CCR Rule Overview
- Next steps in CCR Compliance Timeline
 - Detection Monitoring
 - Error Analysis
 - Natural Groundwater Quality Variation
 - Alternate Source Determination
 - Assessment Monitoring
 - Same as Above
- What is Unknown



Detection Monitoring

Appendix III	Appendix IV	
Boron	Antimony	Lead
Calcium	Arsenic	Lithium
Chloride	Barium	Mercury
Fluoride	Beryllium	Molybdenum
pH	Cadmium	Radium 226+228
Sulfate	Chromium	Selenium
Total Dissolved Solids	Cobalt	Thallium
	Fluoride	

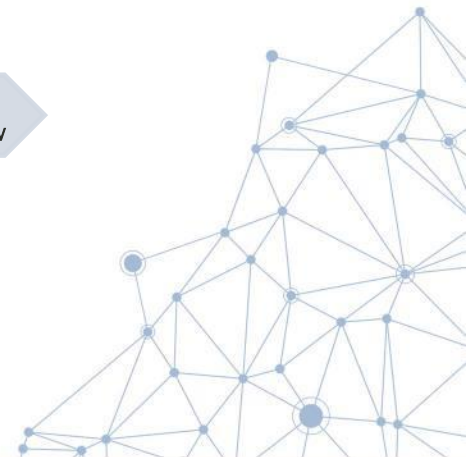
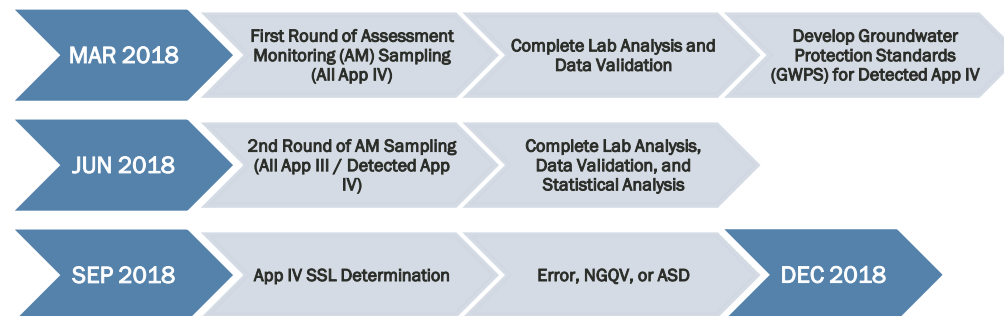
- Minimum semi-annual (2x year) monitoring for Appendix III



Assessment Monitoring-Baseline Data

Appendix III	Appendix IV	
Boron	Antimony	Lead
Calcium	Arsenic	Lithium
Chloride	Barium	Mercury
Fluoride	Beryllium	Molybdenum
pH	Cadmium	Radium 226+228
Sulfate	Chromium	Selenium
Total Dissolved Solids	Cobalt	Thallium
	Fluoride	

- If Appendix III SSI's are confirmed (no errors, natural groundwater variation, or alternate sources), initiate Assessment Monitoring

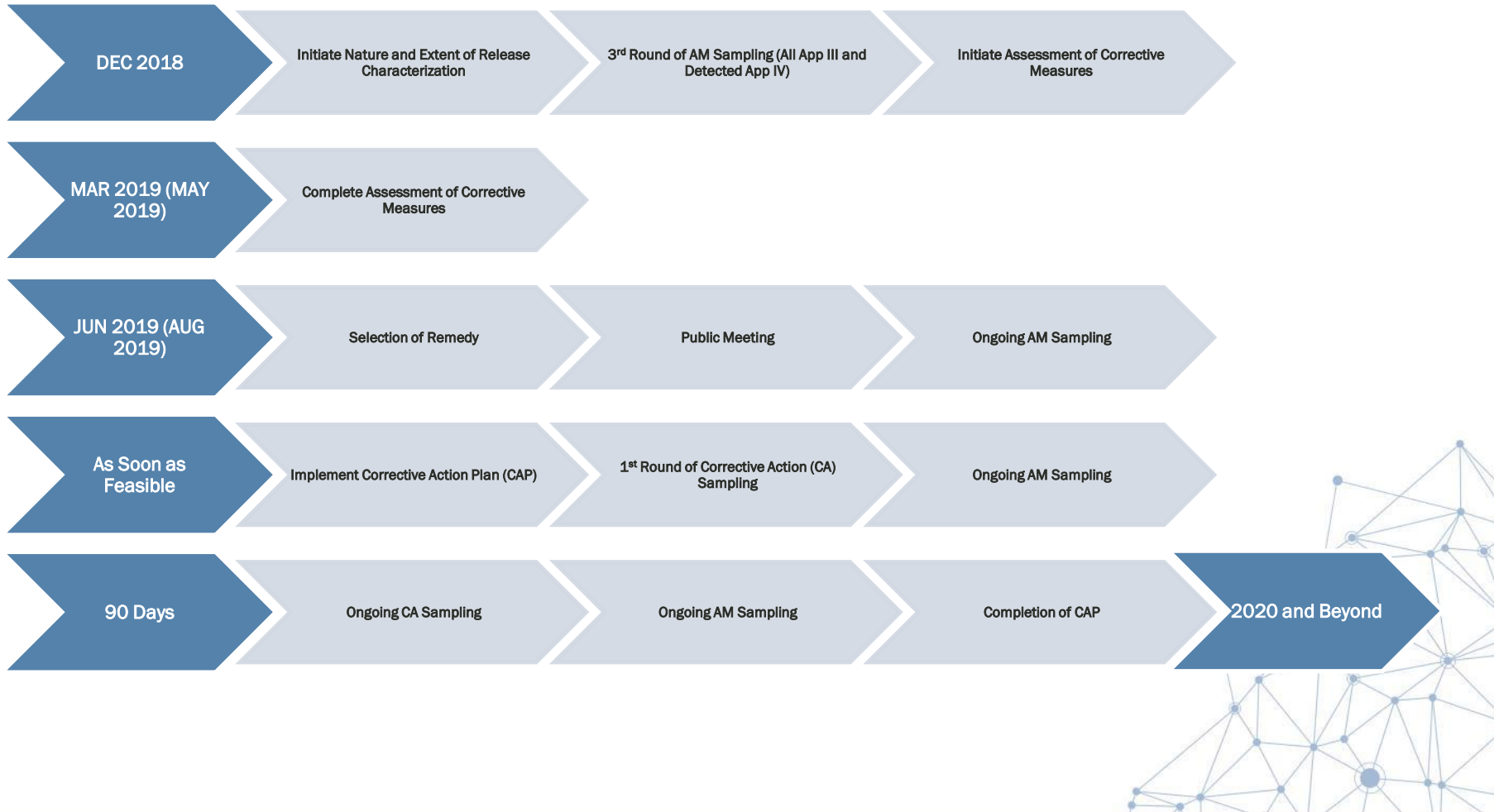


Assessment Monitoring-Baseline Data

- Initially analyze for App IV and establish Groundwater Protection Standards (GWPS) for each App IV constituent
- GWPS = Maximum Contaminant Level (MCL) if one exists, or background concentration, whichever is higher
- Within 90 days, resample for App IV constituents that were previously detected plus App III list
- If Appendix IV constituents are detected at a Statistically Significant Level (SSL) > GWPS, must characterize the nature and extent of the release
- Within 90 days, initiate assessment of corrective measures
- If SSL > GWPS for an unlined impoundment, cease receipt of CCR and initiate closure within 6 months (Ash Pond June 2019)

Appendix III	Appendix IV	
Boron	Antimony	Lead
Calcium	Arsenic	Lithium
Chloride	Barium	Mercury
Fluoride	Beryllium	Molybdenum
pH	Cadmium	Radium 226+228
Sulfate	Chromium	Selenium
Total Dissolved Solids	Cobalt	Thallium
	Fluoride	

Corrective Measures Assessment & Selection of Remedy



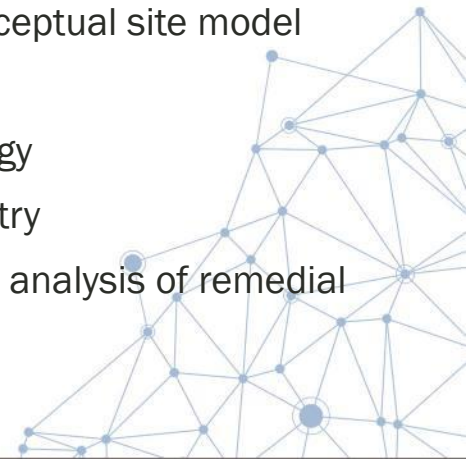
Corrective Measures Assessment & Selection of Remedy

- **Corrective Measures Assessment Goals:**

- Prevent further releases
- Remediate any releases
- Restore affected area to original conditions

- **Step 1 – Collect and Evaluate Additional Data (Remedial Investigation/RFI)**

- Define nature and extent of contamination
 - Additional monitoring wells, dye traces
 - Additional sampling for broad suite of analytes
 - Update the conceptual site model
 - Geology
 - Hydrogeology
 - Geochemistry
 - Data to support analysis of remedial alternatives



Corrective Measures Assessment & Selection of Remedy

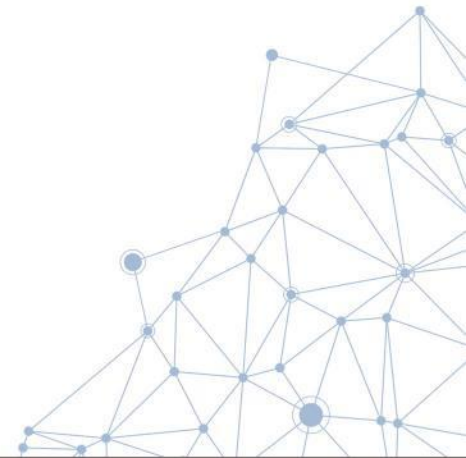
- **Step 2 – Evaluate a Range of Corrective Measures (Feasibility Study/CMS)**
 - Select feasible remedial approach
 - Addresses the contaminants
 - Implementable
 - Cost-effective
- **Step 3 – Select preferred approach (Proposed Plan)**
 - Detailed analysis of selected approach
 - Regulator approval
 - Public meeting
- **Step 4 – Finalize preferred approach (Record of Decision/Statement of Basis)**
 - Final adjustments to preferred approach
 - Schedule for design and implementation
- **Step 5 – Detailed design (Remedial Action Work Plan/Design)**
 - Possible Pre-design data collection
 - Define performance evaluation criteria
- **Step 6 – Design Implementation (Remedial Action/CMI)**
 - Remedy construction/installation
 - Remedy operation and maintenance



Corrective Measures Assessment & Selection of Remedy

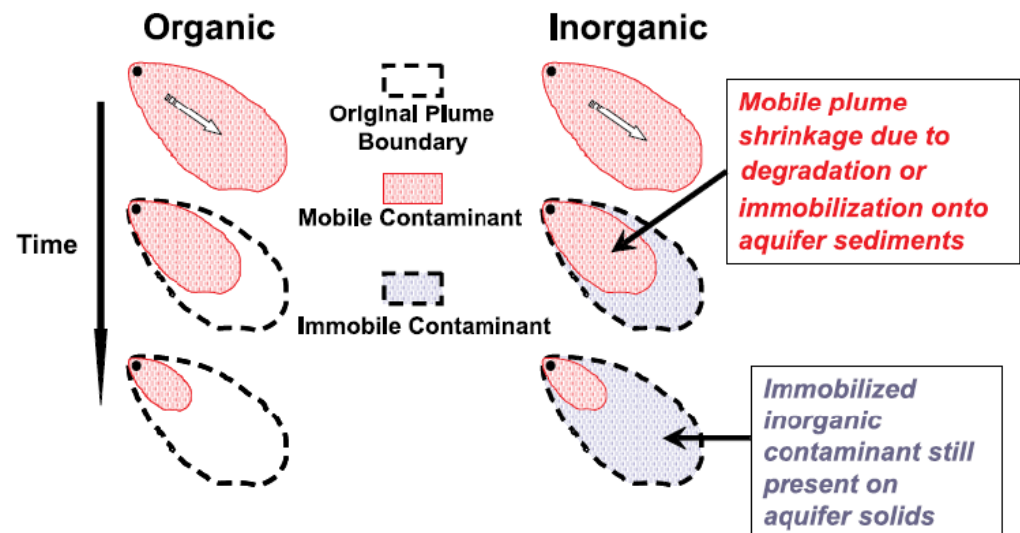
- **Options**

- Pump and Treat (existing water treatment systems)
- Monitored Natural Attenuation (MNA)
- In-Situ Treatments (i.e., Sorption/Precipitation)
- Permeable Reactive Barriers
- Cut-Off Walls (in combinations)



MNA – Application to Inorganics


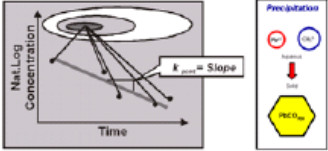
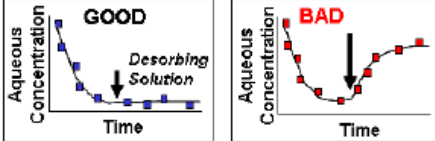
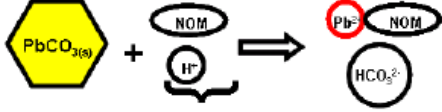
- **Inorganic MNA applicable to the Appendix IV metals**
- MNA relies on physical and chemical processes in the aquifer to address mobile contaminant
- MNA as a remedy normally requires:
 - Source control
 - Detailed conceptual model to predict behavior
 - Long-term monitoring with periodic updates of the conceptual model

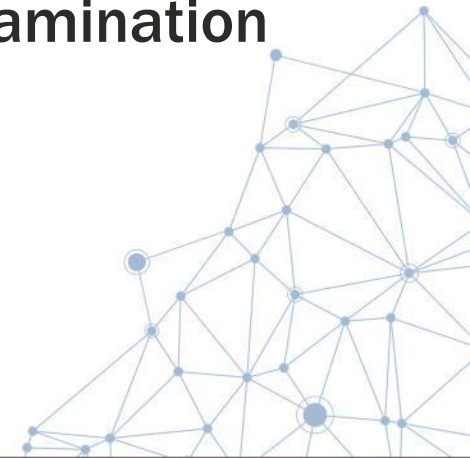


Source: ITRC
2010

MNA and Appendix IV Inorganics

- Data to evaluate MNA as corrective action collected as part of investigation to define nature and extent of contamination

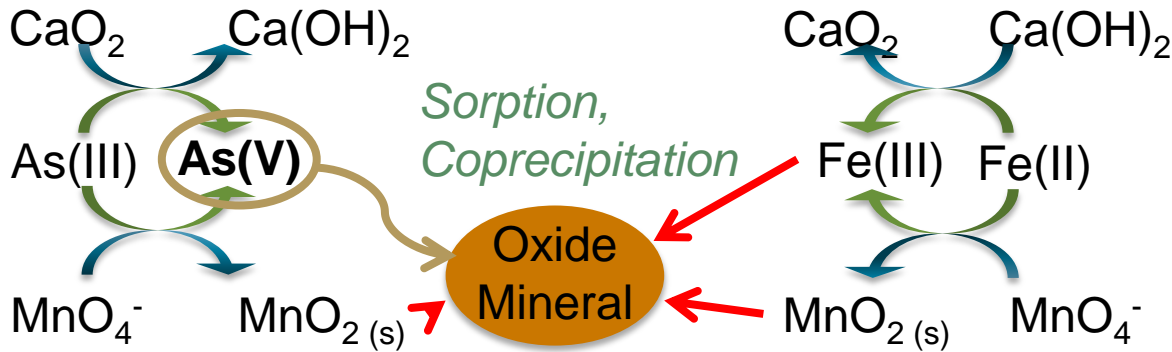
Tier 1	 <p style="text-align: center;"><i>Plume Evolution</i></p>	Tier I: Plume Stable or Shrinking
Tier 2		Tier II: Identification of Attenuation Mechanisms
Tier 3		Tier III: Attenuation Ability and Capacity
Tier 4	 <p style="text-align: center;"><i>Performance Monitoring Parameters</i></p>	Tier IV: Long Term M&M



Manipulate Aquifer Geochemistry

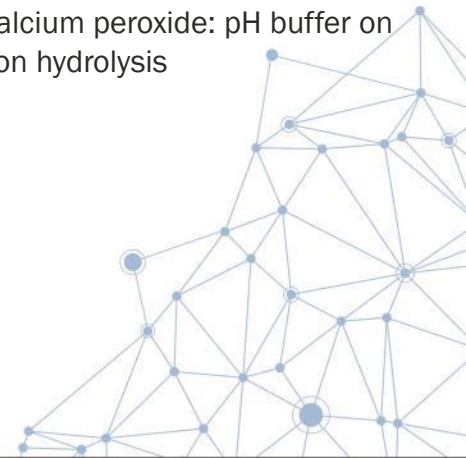
Arsenic Oxidation

Iron Oxidation



- **Approach: Oxidize As (III) to As(V), Increase sorptive capacity**

- Oxidants under study: Permanganate, Calcium Peroxide
- Manganese oxides: Increase sorptive capacity
- Ferrous sulfate oxidation: Increase sorptive capacity
- Calcium peroxide: pH buffer on iron hydrolysis



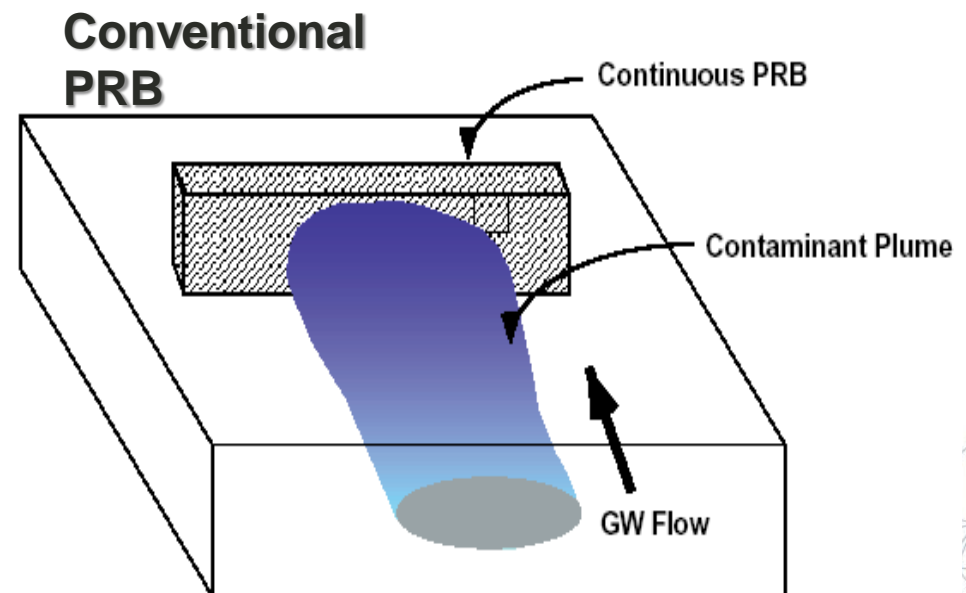
Manipulate Aquifer Geochemistry

- Can you manipulate the aquifer to precipitate Appendix IV metals from groundwater?
- Metal phosphates, sulfates, sulfides and hydroxides can be insoluble
 - Permeable reactive barrier design could incorporate sources of anions
 - Aquifer chemistry may be manipulated to precipitate metals (e.g., modification of in situ bioreactor)

Appendix IV Metal	K_{sp}			
	Phosphate	Sulphate	Sulfide	Hydroxide
Antimony			1.6×10^{-93}	
Arsenic	Anionic			
Barium	3.4×10^{-23}	1.08×10^{-10}		2.55×10^{-4}
Beryllium				6.92×10^{-22}
Cadmium	2.53×10^{-33}		1×10^{-27}	7.2×10^{-15}
Chromium				3×10^{-29}
Cobalt	2.05×10^{-35}		5×10^{-22}	5.92×10^{-15}
Lead		2.53×10^{-8}	3×10^{-28}	1.43×10^{-20}
Lithium	2.37×10^{-11}			
Mercury		6.5×10^{-7}	2×10^{-53}	3.6×10^{-26}
Molybdenum	Anionic			
Radium		3.66×10^{-11}		
Selenium	Anionic			
Thallium			6×10^{-22}	1.68×10^{-44}

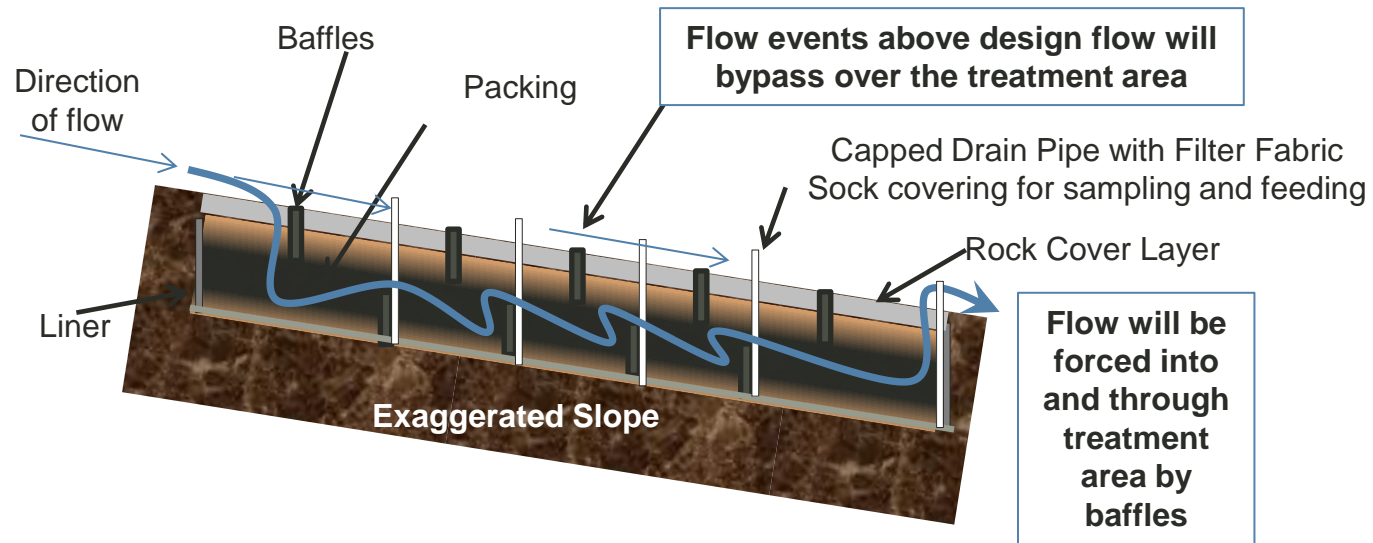
Permeable Reactive Barrier

- Reactive media in the PRB tailored to the contaminants of concern, e.g.,
 - Alumina to adsorb As
 - Apatite (phosphate) to precipitate lithium
 - Iron filings to sorb arsenic and molybdenum oxyanions
 - Sulfate source to precipitate radium

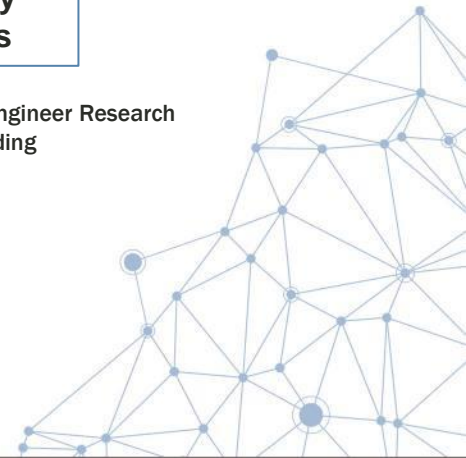


Source: USEPA

Field Design of Packed Bed



Source: US Army Corps of Engineers, Engineer Research and Development Center – Patent Pending



Pilot Scale Implementation

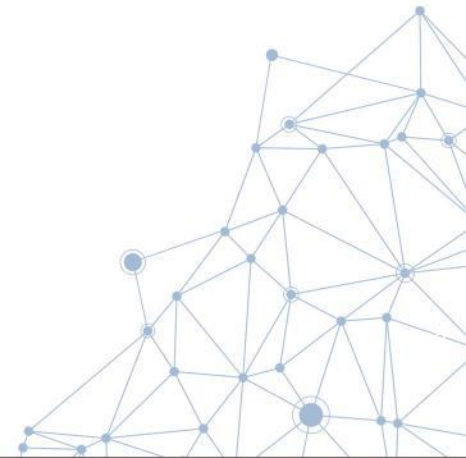


Sampling and Data Logging Unit



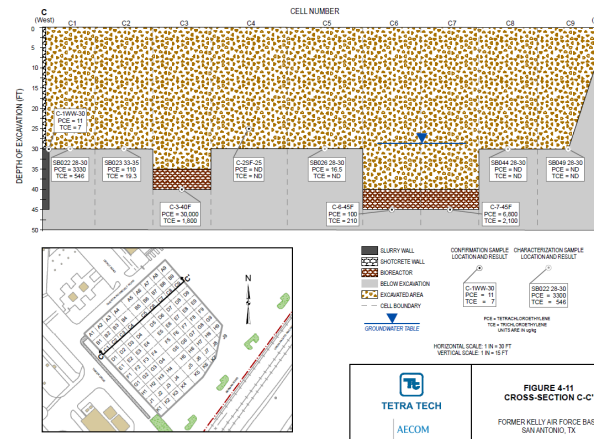
Permeable Reactive Barrier Installation

- Long Stick Excavation
 - Slurry Trench
 - Bench Setting
- One-Pass Trencher

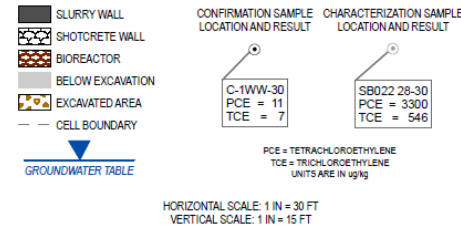
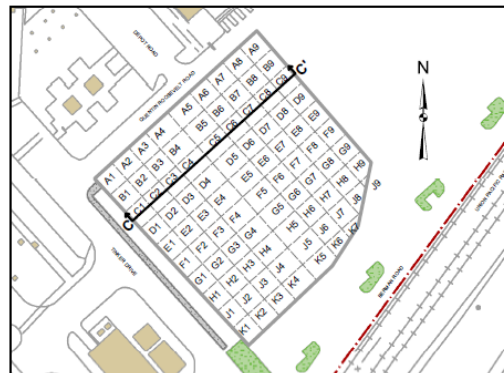
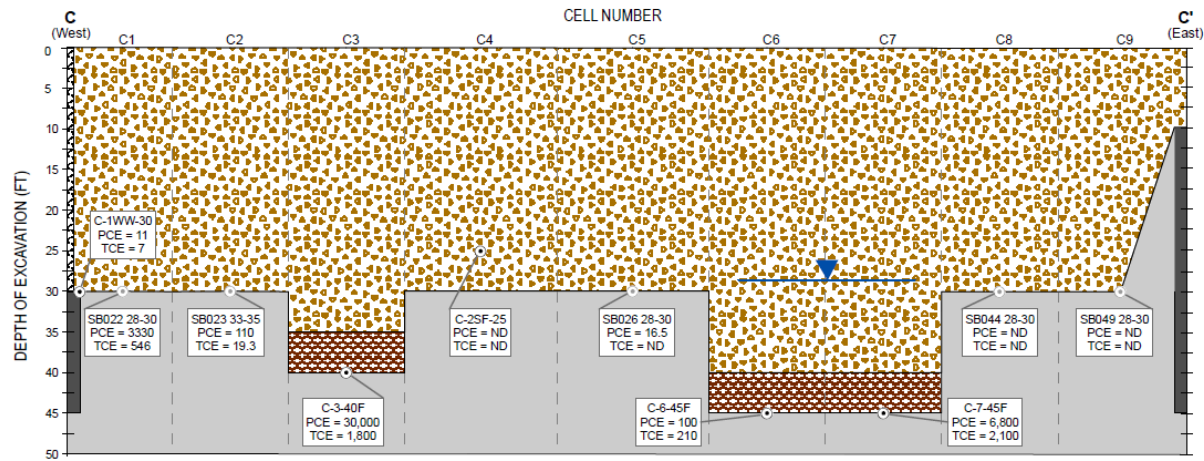


In Situ Bioreactors

- Engineered subsurface mechanism to control the groundwater chemistry
 - Provides an initial “jolt” to the system
 - Mechanism for recharging the feed system
 - Biotic or abiotic actions can be stimulated



Case Study I – In Situ Bioreactors



**FIGURE 4-11
CROSS-SECTION C-C'**

FORMER KELLY AIR FORCE BASE
SAN ANTONIO, TX

Summary and Questions

- Sites often present physical limitation on remedial approaches (e.g., proximity to rivers)
- Sites often have extensive experience with conventional water treatment systems – pump and treat may be an attractive option
- CCR-related metals can be addressed with conventional remedial approaches
- CCR-related metals may be amenable to innovative in situ approaches
- Questions

